## CHEM 344: Synthesis of 4'-propoxyacetophenone under phase-transfer conditions

organic phase 
$$\bigoplus_{Bu_4N} \bigoplus_{Bu_4N} \bigoplus_{Bu_4N}$$

- 1) NaOH deprotonates 4'-hydroxyacetophenone to give the phenolate anion.
- 2) The  $Bu_4N^+$  cation pairs up with the phenolate anion.
- 3) The TBA/phenolate species transfers into the organic phase.
- 4) The phenolate anion undergoes an  $S_N2$  reaction with 1-bromopropane.
- 5) Bu<sub>4</sub>NBr is regenerated and transfers back into the aqueous phase.

Note that OH- does not initially react with 1-bromopropane due to phase separation.