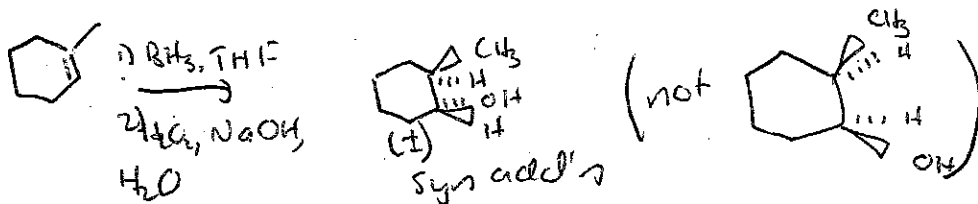
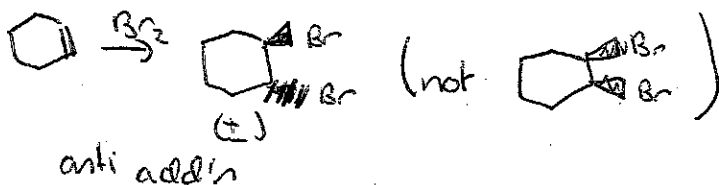


Submit a Single-sided Copy to the Office

DO NOT STAPLE

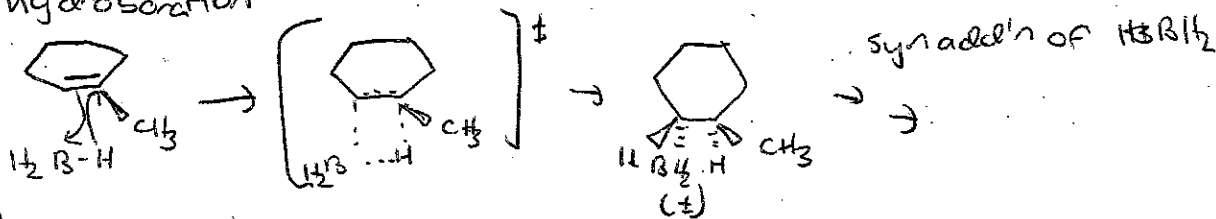
Recall: stereochemical consequences of rxns.

ex

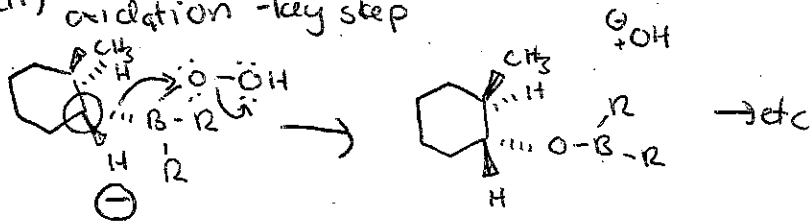


Mechanistic rationale:

(i) hydroboration

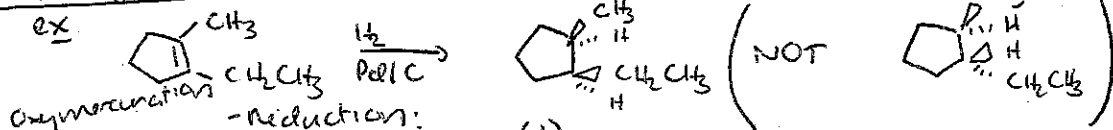


(ii) oxidation - key step

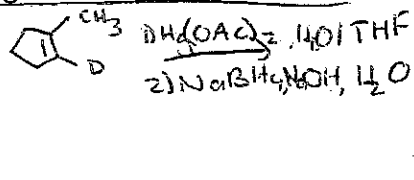


Thus, H₂O₂ add in syn fashion

Hydrogenation - syn addition of H₂



Asymmetric reduction:



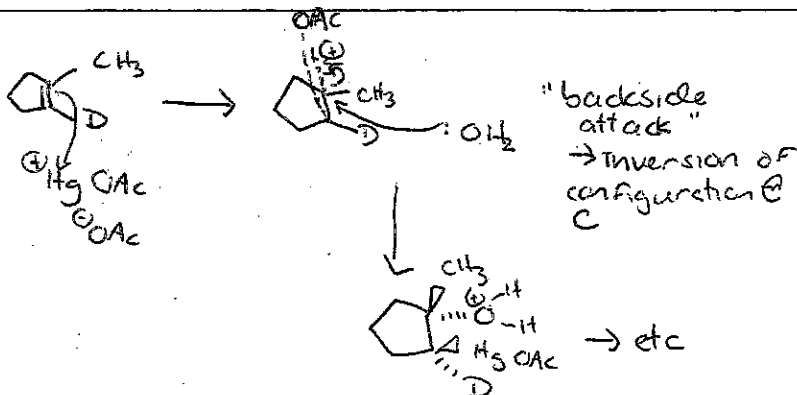
both syn & anti addition products
 \Rightarrow not stereoselective

Submit a *Single-sided Copy* to the Office

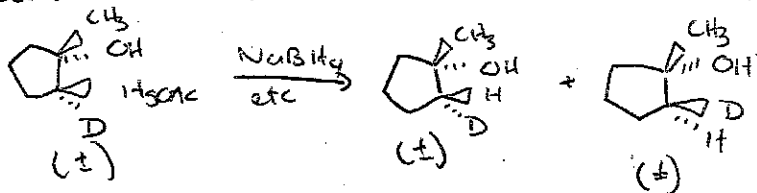
DO NOT STAPLE

Mechanistic Rationales:

Oxymercuration - anti selectivity



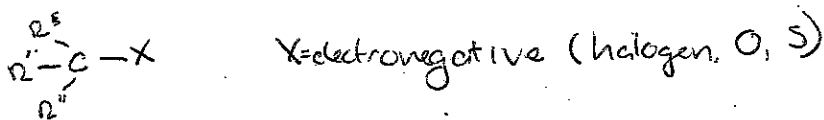
Reduction - non-stereoselective



Chapter 8 - Alkyl Halides, alcohols, ethers, thiols & thioethers

Rec. Problems: 2, 4, 6, 12, 14, 20, 23, 24, 26-32, 33, 34, 39-44, 46, 48, 49, 52, 53, 55-62
not p 6

Common theme:



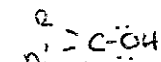
R's: H, alkyl

Alkyl halides-

F, Cl, Br, I

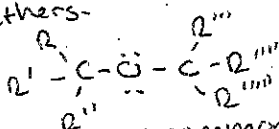
ex $CH_3CH_2CH_2Br$ "ethyl bromide"

Alcohols-



ex CH_3
 $H_3C - C - OH$ "t-butanol"
 $|$
 CH_3

Ethers-



ex THF common organic solvents CH_3
diethyl ether $H_3C - C - O - CH_3$
 $|$
 CH_3 methyl, t-butyl ether

Course 343-2

Lecturer Gellman

Day Friday

Date 10/26/12

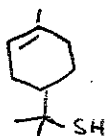
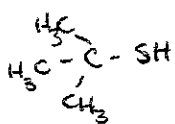
Notes Taken By Markus Hager

Total # of Pages 3

Submit a *Single-sided Copy* to the Office

DO NOT STAPLE

Thiols - Analogues of alcohols



Thioethers



Because aprotic halides, alcohols & ethers are common organic solvents, we must understand their physical properties

key distinctions

non-polar vs polar
"apolar"