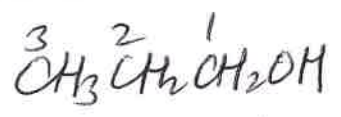


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Are H's (or other chemical groups) w/i a molecule equivalent to one another?

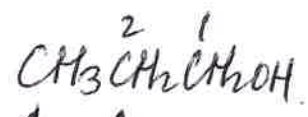
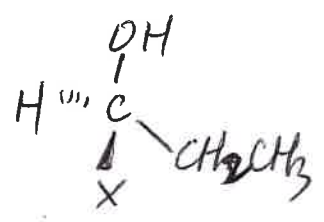
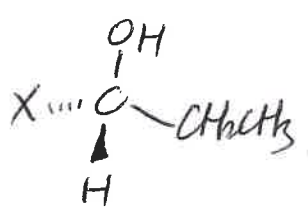
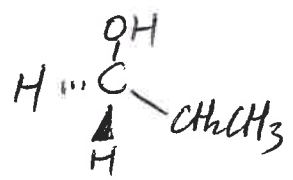
Consider:



H's on C₁, C₂, C₃ different

What about the H's on C₁!

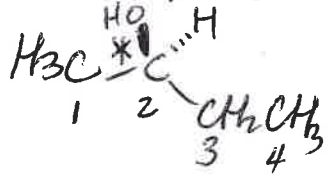
C-enantiotopic: If substitute one for something else, and then the other, you obtain enantiomers.



↑ ↑
enantiotopic

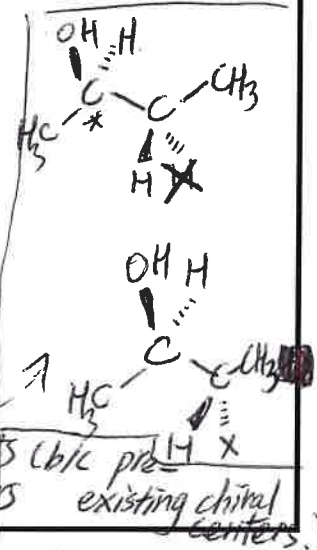
homotopic: substituting any gives same product.

(R)-2-butanol



H's on C₁, C₄ = homotopic

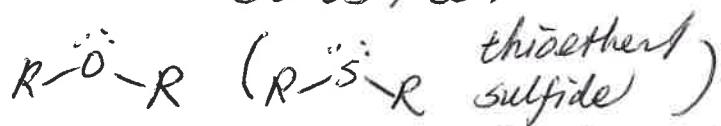
" " C₃ = diastereotopic: replacement products (w/ existing chiral centers) produce diastereomers



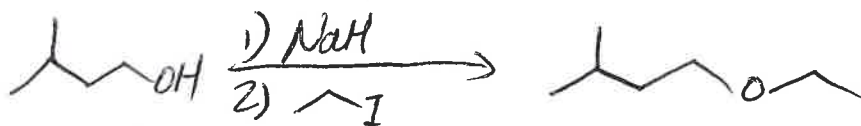
Course Chem 343 Lecturer Gollman
 Day Wednesday Date 11-18-15
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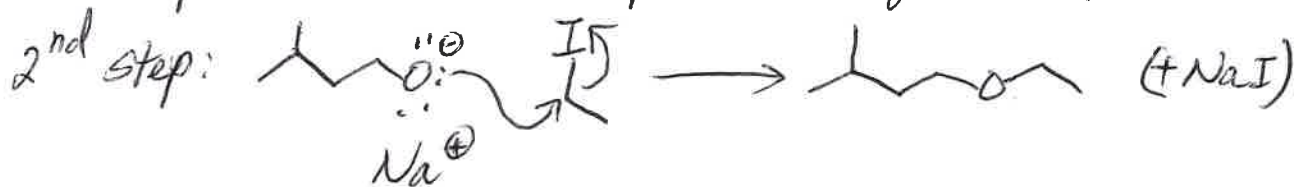
CH 11 → Ethers & related compounds
 problems: 1-25, 32, 34, 36, 39, 40 a,b,d, 44, 45 c-j, 46, 48,
 50-63, 65, 69-71, 77, 79-81.



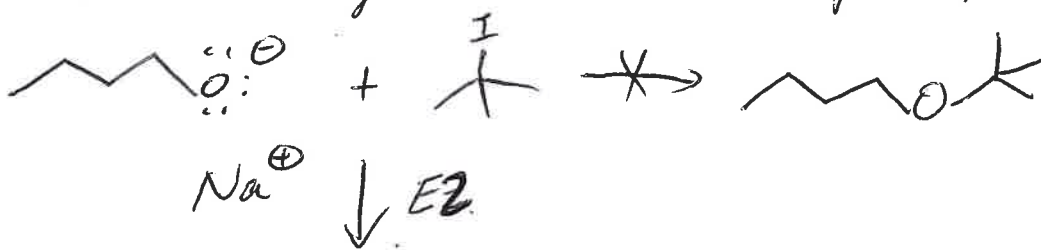
1) Williamson ether synthesis: (S_N2 rxn of alkoxides)



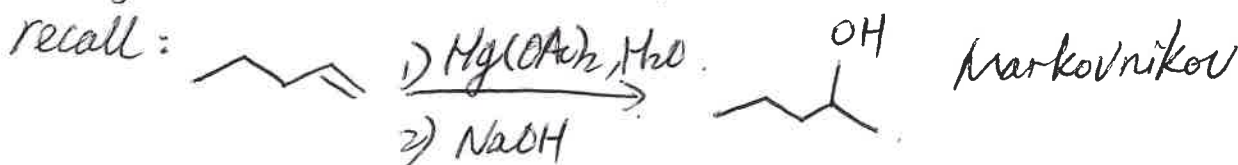
1st step: Quantitative Deprotonation of alcohol.



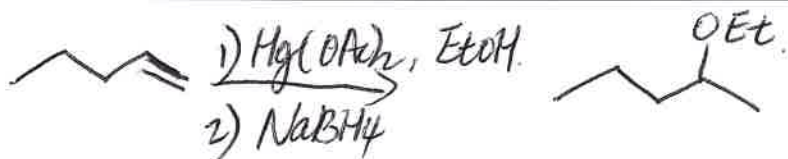
Limitation: alkyl halide must be good for S_N2 . (methyl, 1°)



2) Alkoxymercuration-reduction



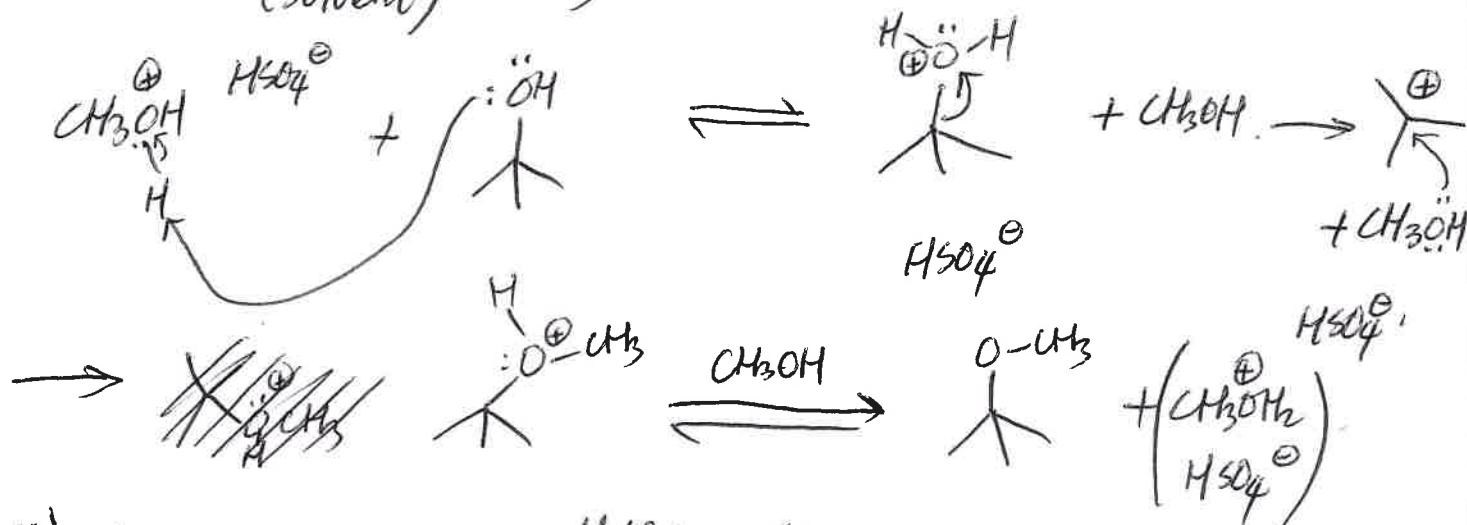
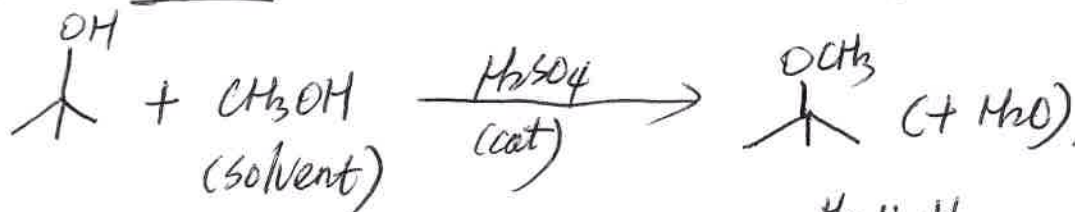
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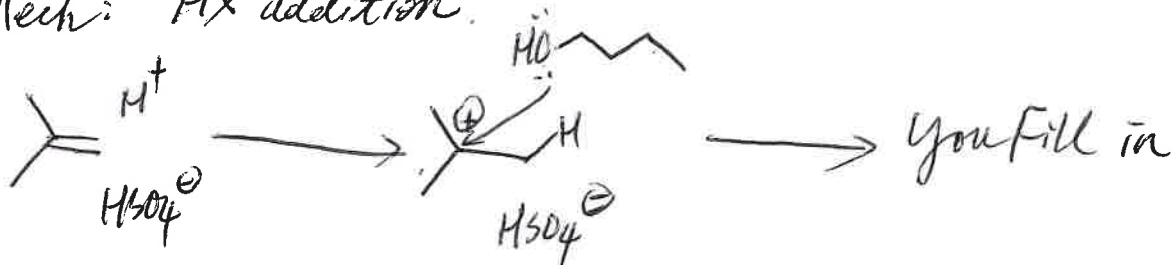
Limitations: simple alcohols, must be used as solvent.

3) Acid catalyzed rxns from alcohols or alkenes

SN1 * Most useful involve rxns of 3° C⁺ w/ an alcohol.



Mech: HX addition.



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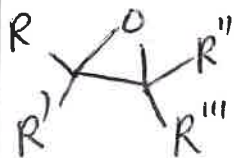
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Ethers can be cyclic



THF

Epoxides = special class of cyclic ethers w/ high importance
(b/c of their reactivity)



reactivity: ring strain