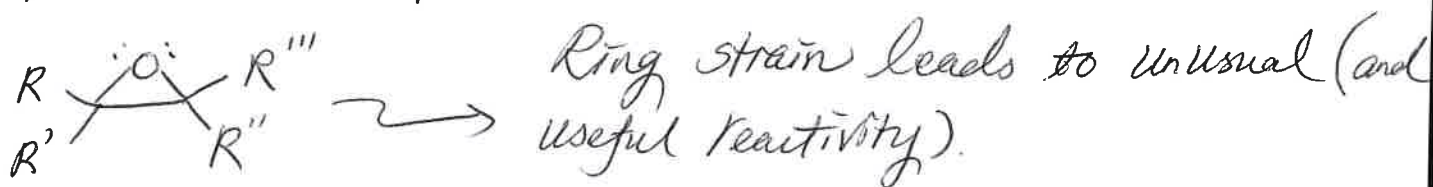


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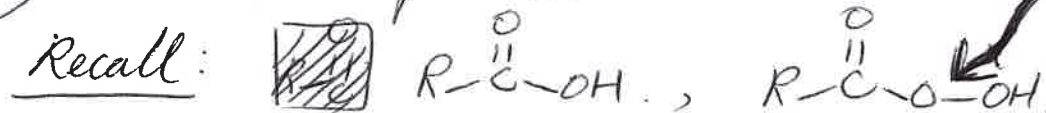
Recall Rxns of Ethers...

Special case. Epoxides.



Synthesis of Epoxides

1) Alkene + "peracid"

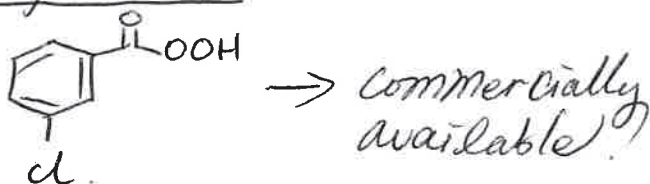


Note: Weak bond (2 electronegative atoms...)

(Recall: peroxide = HO-OH)

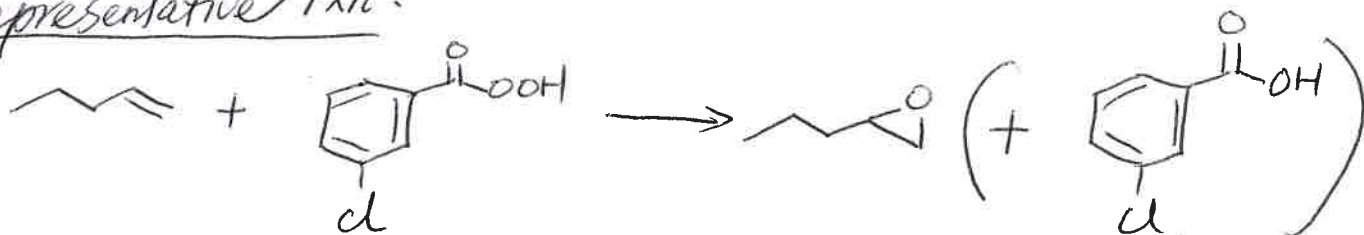
"per-carboxylic acid" ("peracid")

Specific case



meta-chloroperoxybenzoic acid (mCPBA)

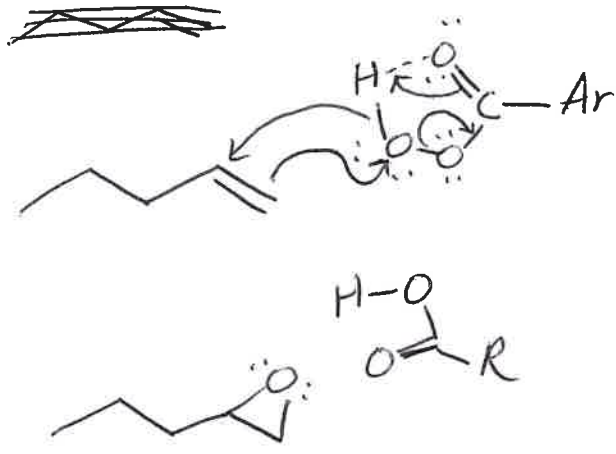
Representative rxn:



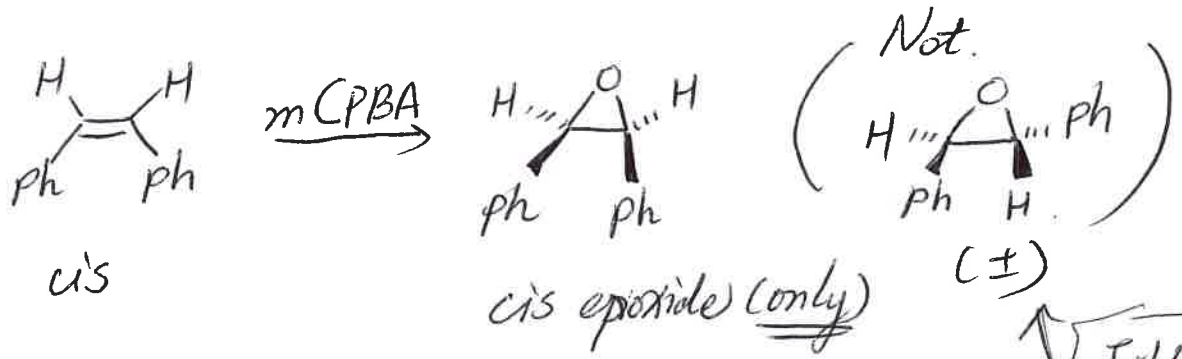
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Recall: Alkene + Br<sub>2</sub>/Ar

Mech:

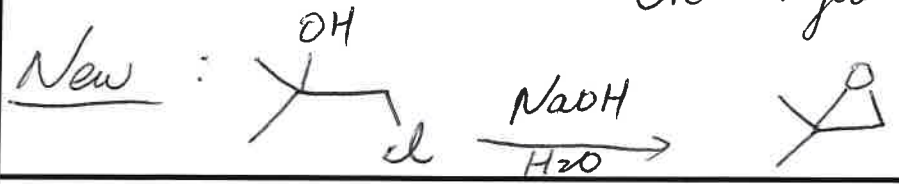
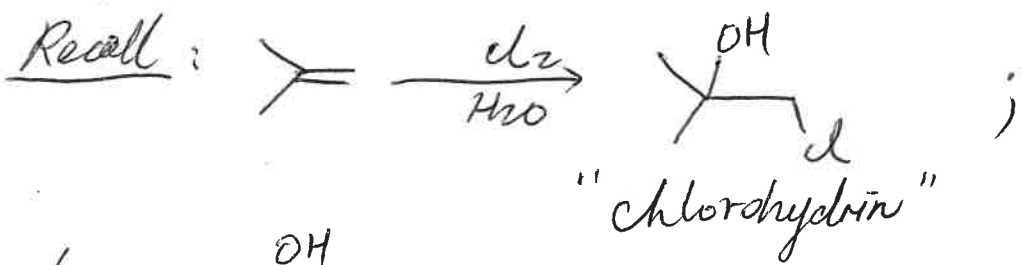


Stereochemical evidence supports hypothesis of concerted mech.



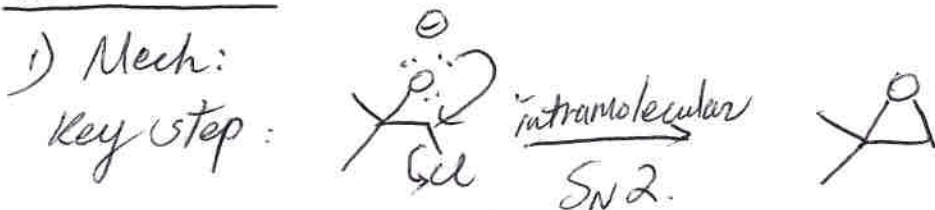
Fill in how to get this product.

2) Cyclization of halohydrins



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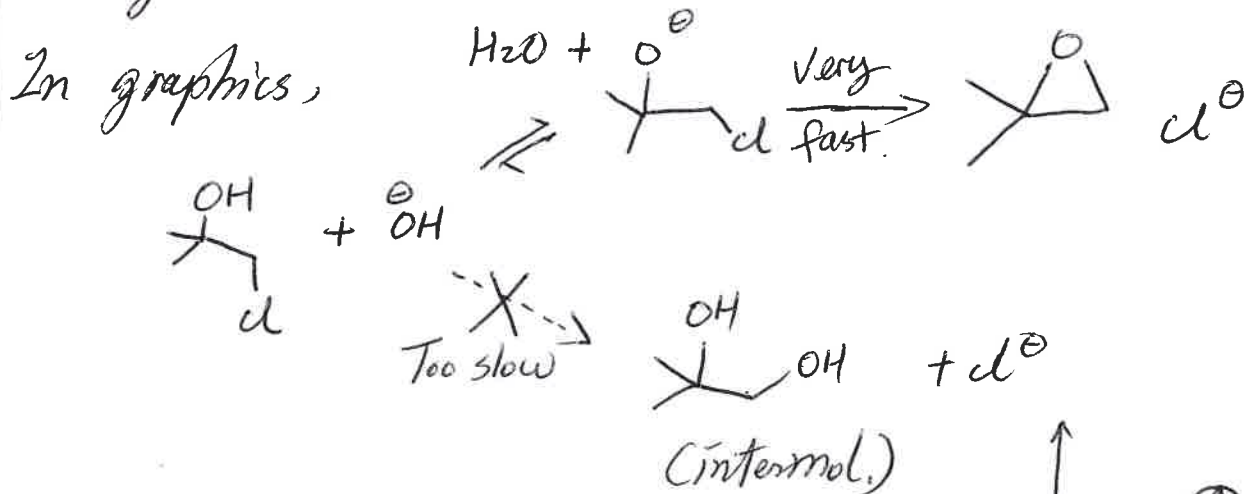
Comments:



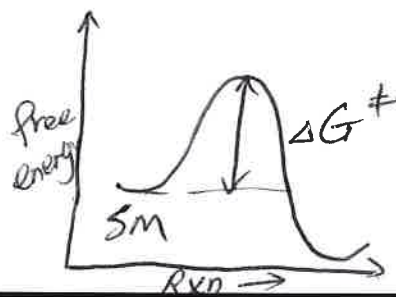
2) Recall previous discussions of SN2 rxn of alkoxides w/ alkyl halides; need for quant. deprotonation to generate alkoxide (NaH).

Why do we not worry about quant. deprotonation in this case?

→ Intramolecular rxns have intrinsic advantages over intermolecular rxns, esp. when forming 3-6 membered rings.



Reactivity  $\approx$  energy of activation  
 $\Delta G^\ddagger = \Delta H^\ddagger - T\Delta S^\ddagger$



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$$\Delta G^\ddagger = \Delta H^\ddagger - T\Delta S^\ddagger$$

$\downarrow$  enthalpy of activation  
 $\rightarrow$  entropy of activation

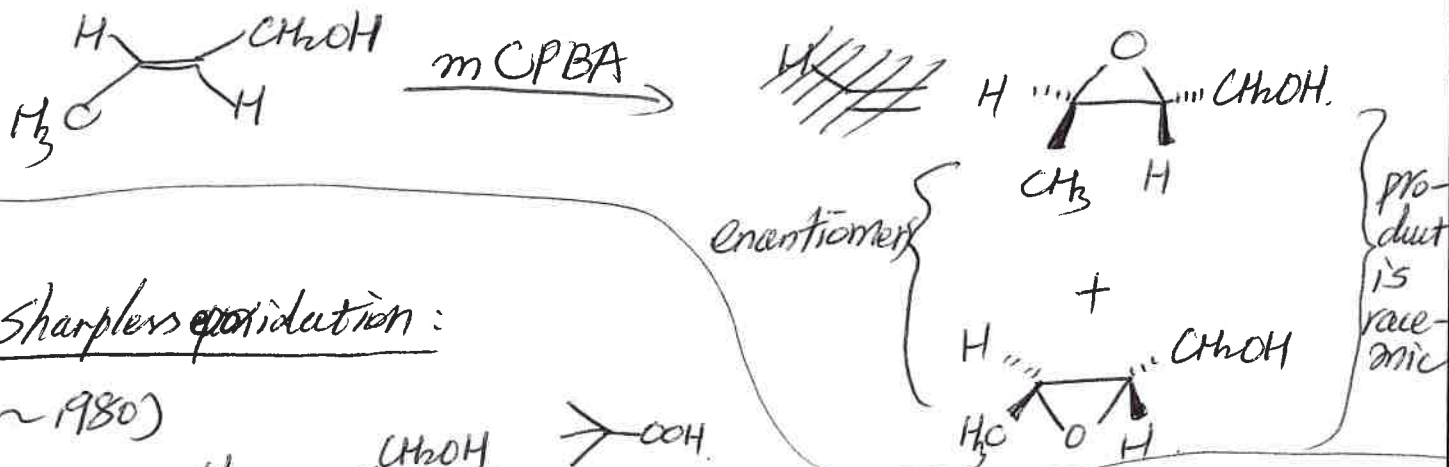
Intramolecular rxns are much less entropically unfavorable than intermolecular.

Stereochemistry in intramolecular rxn....

3) [§ 11.11] Asymmetric ~~epoxide~~ epoxide synthesis.

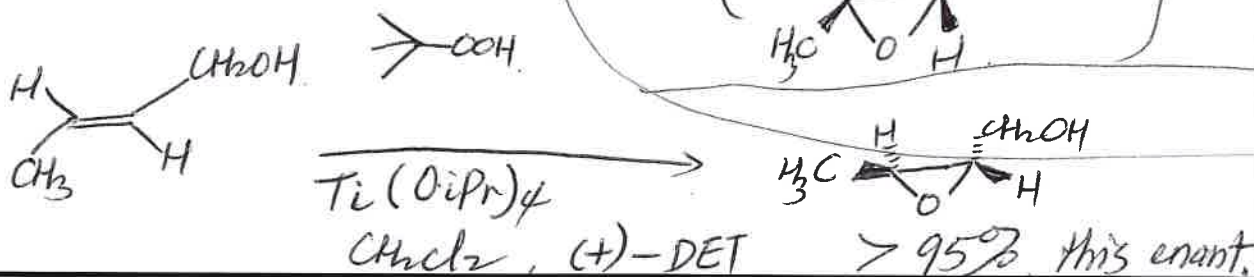
(i.e. synthesis of chiral epoxides that favors one enantiomer over the other).

Consider:



Sharpless epoxidation:

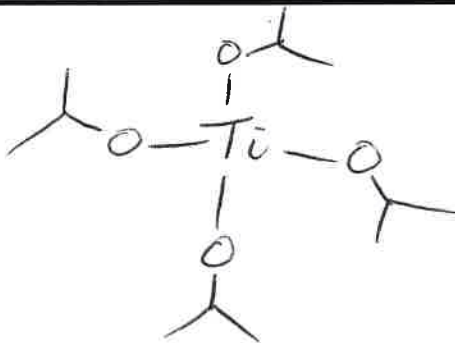
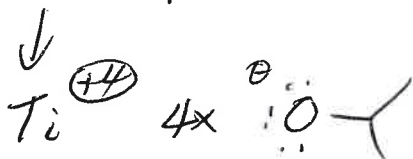
(~1980)





Course Chem343 Lecturer Gellman  
Day Friday Date 11-20-15  
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