Final Exam Chemistry 345 Professor Gellman 13 May 2010

Last Name Answer

First Name __

General Instructions:

- (i) Use scratch paper at back of exam to work out answers; final answers must be recorded at the proper place on the exam itself for credit.
- (ii) Print your name on each page.
- 1. (39 points) Show the major product or products expected from each reaction:

(b)
$$Ph_3P = \Delta$$

-- cont. on next page --

1	(co	nt)
1.	100	

Name ____

- 1) 1 equiv. NaOH, H₂O room temperature
- 2) HCI (neutralize)

- 1) 1 equiv. CH₃CH₂ONa CH₃CH₂OH
- 2) HCI (neutralize)

[Note: In the IR region 1700 - 1800 cm⁻¹, the starting material has a single strong signal at 1745 cm⁻¹, but the product has two signals, at 1745 cm⁻¹ and 1725 cm⁻¹.]

2. (45 points) Show the reagents and other organic molecules required to convert the starting material to the indicated product. Be sure to differentiate clearly between distinct steps, by using "1)", "2)", etc.

(cont. on next page)

Name		

2. (cont.)

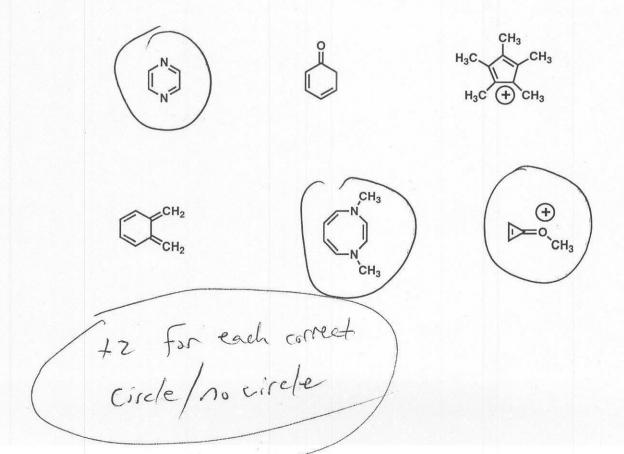
(f)
$$H_3C$$
 O_2 O_3 O_3 O_4 O_4 O_3 O_4 O_4 O_4 O_5 O_5

Name	

3. (8 points) For each set of compounds below, rank the molecules, left to right, from lowest pK_a to highest pK_a (i.e., your answer should have the form X < Y < Z).

(a) F OH OH OH OH OH

4. (12 points) Among the species shown below, CIRCLE those that you would expect to benefit from aromatic stabilization.



Name		

5. (10 points) The cation shown on the left below would be expected to react with cyclopentadiene to form the two isomeric bicyclic cations shown on the right (these intermediate would then go on to form more stable species). Provide a molecular orbital rationale for the expectation that these specific bicyclic cations would form (and not isomers with *trans* methyl groups).

Stable products

Reaction expected is a cycloaddition involving the allylic Cation LUMO and the diene HOMO; Storeschemisty arises becomes the seaction is concerted. Allylic Calin LVMO Both points of confact Diene Homo o-bonds) correspond to bondie

internations.

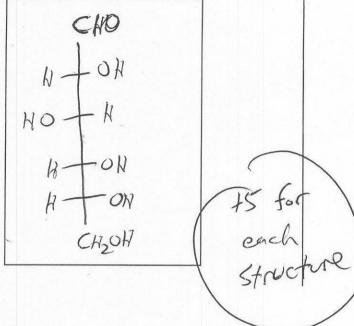
Started at exam: "Be sure to draw major resonance stratures." 6. (18 points) Draw a mechanism (curved arrows) for the reaction shown below. + 2 for each set (a) of mechanishe grows (19 total) # for each 061 pes. structure of intermediate (44 total) + AN + 1 for each intermediate structure (to total) CH2CH3

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7. (20 points) When the aldehyde shown below is subjected to the sequential reaction conditions indicated, A is formed. Further reactions, as shown, generate B and then a final product that turns out to have two chromatographically separable and isomeric components, C and C'. Give the structures of A, B, C and C'.

8. (20 points)

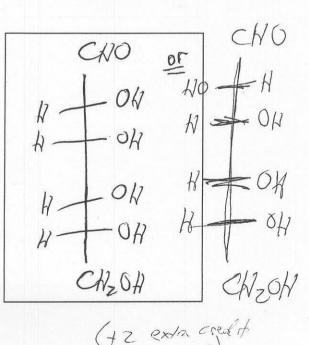
- (a) Draw the Fischer projection of D-glucose (open chain form; not cyclic).
- (b) Mannose is the 2-epimer of glucose. Draw the Fischer projection of L-mannose (open chain form; not cyclic).



- TH CH20H
- (c) Compound X is an aldohexose. When X is treated with 3 equivalents of PhNHNH2 in acetic acid, the osazone formed is different from the osazone obtained from D-glucose or D-mannose (these two give the same osazone). However, when X is subjected to two cycles of the Wohl degradation process, the resulting aldotetrose, D-erythrose, is the same as the aldotetrose generated from two cycles of Wohl degradation of D-glucose or D-mannose.

Draw the Fischer projection of D-erythrose.

Draw the Fischer projection of X.



(+2 extra credit

Name	

9. (12 points) When peptide Z is treated with CNBr, three short fragments are generated, as shown below. When peptide Z is instead treated with the enzyme trypsin, a different set of three short fragments is generated, as shown. Based on this information, what is peptide Z?

Fragments from degradation with CNBr:

Fragments from degradation with trypsin:

SCH3

NH

NH

SCH3

NH

NH

SCH3

SCH3

SCH3

Ala-Lys-Met-Ala-Arg-Met-Phe-Val +115 Am coprect order but missing I residue

Name	

10. (16 points) Propose a synthesis of the target molecule shown below from the indicated starting material. You may use other reagents that contain no more than one carbon atom.