

Course Chem 345 Lecturer Prof. Gellman
Day Friday Date 4/29/15
Notes Taken By Cassie Jarvis Total # of Pages 5

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Recall: (Dr. MacDonald = lecturer today)

Chp. 27: All problems are recommended, but some towards the end are tricky

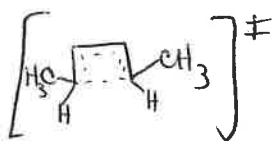
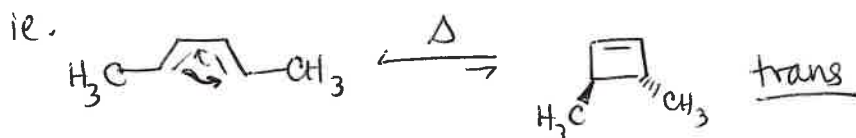
- Pericyclic Reactions \Rightarrow These reactions feature a concerted mechanism with a cyclic array of partial bonding at the TS (transition state).

ie. Diels-Alder cycloaddition:



- Electrocyclic Reactions (to start with)

* Definition: unimolecular reactions that result in "trading" a π bond for a σ bond or vice versa



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* Problem :

= curved arrows do not explain the stereospecificity of reactions

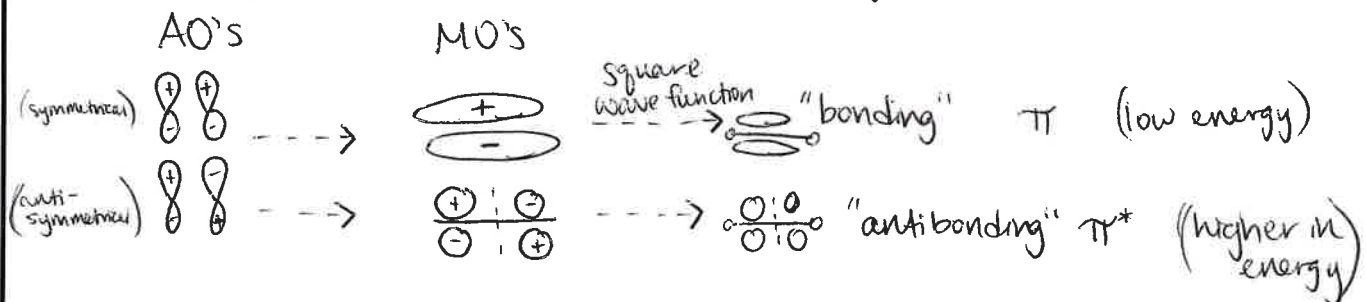
↳ must look @ p-orbitals to understand

* Prelude :

p-orbital

 or  ⇒ the region of space is described by a wave function (ie. probability of finding an e- w/in this region is the square of the wave function)

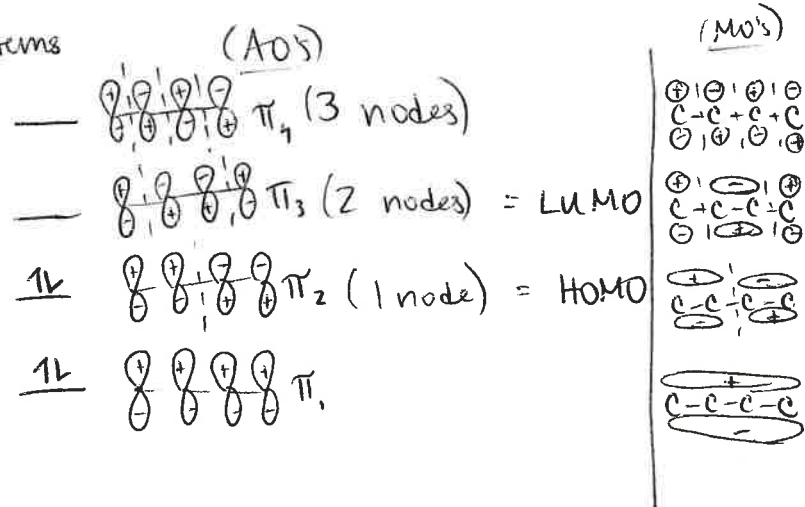
* Two ways to combine p orbitals from adjacent atoms



- Apply to larger π systems
 (ie. )

4 Carbon ZP orbitals → 4 x CZP

E↑



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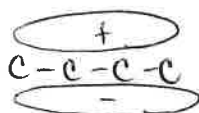
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+ All of this allows us to assign a feature of π MO's that was previously invisible

- Consider:



or



\Rightarrow symmetric (MO sign orientation identical for two ends)



or



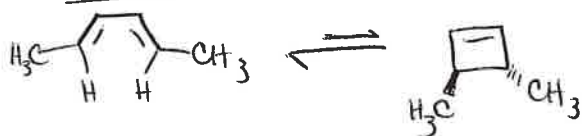
\Rightarrow anti symmetric (signs of end not same)

(assign $\pi_3 + \pi_4$ yourself...)

- Note:

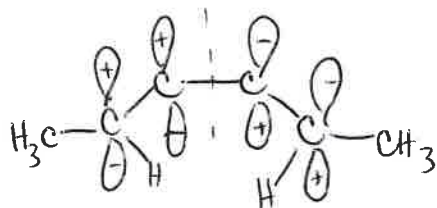
- 1) Every π MO in any organic molecule is either symmetric or anti-symmetric
- 2) For a given conjugated alkene (π system), S vs. A alternates as we move up the energy state

* Recall:



How do we explain stereochem?

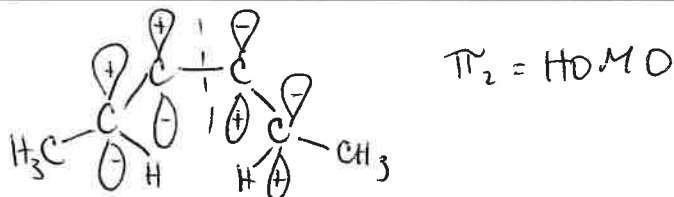
\Rightarrow Draw HOMO on top of molecule



$\pi_2 = \text{HOMO}$

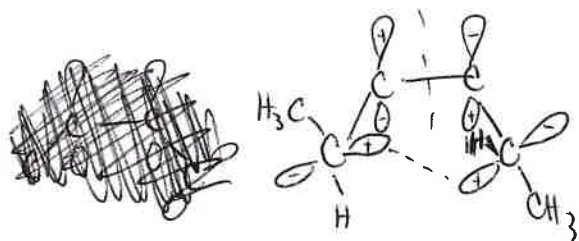
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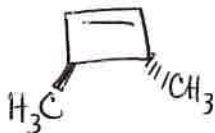


if ends rotate
in the same
direction
= "conrotatory"

if ends rotate
in opposite
direction
= "disrotatory"



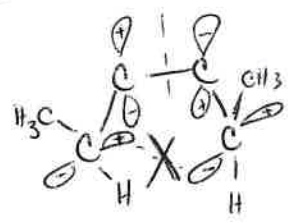
= both ends rotate \curvearrowright (can also rotate both \curvearrowleft for conrotatory)
 \hookrightarrow conrotatory
 = leads to bonding interaction



"conrotatory ring closure" allowed
 = how get trans stereochemistry

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= one rotates ↻ + other ↻
↳ disrotatory
= leads to antibonding



cis disrotatory ring closure forbidden