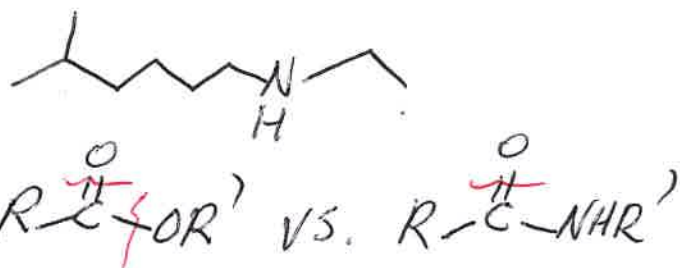
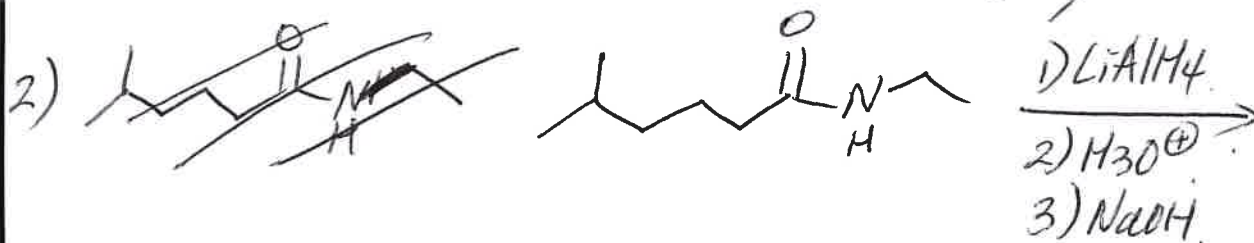
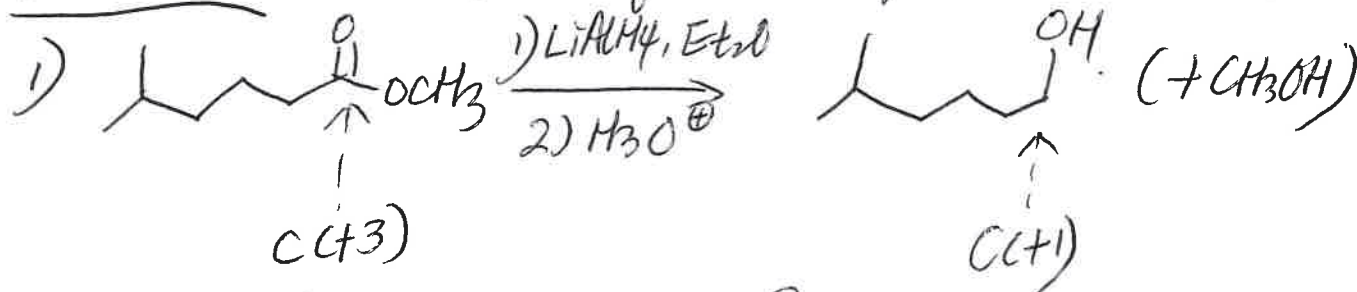


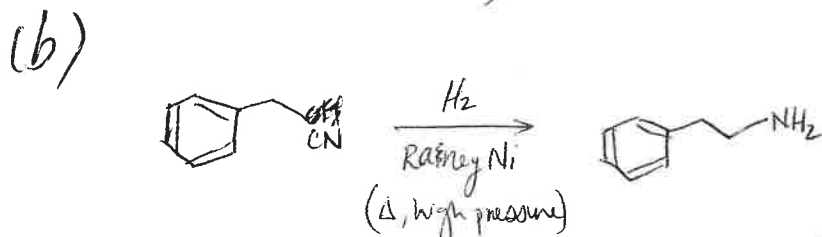
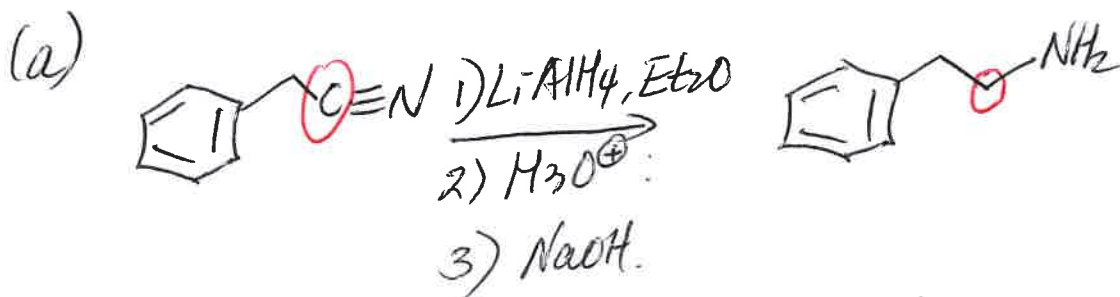
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Recall: Reduction of Carboxylic Acid Derivatives



3) Nitrile redn, to 1° amine.



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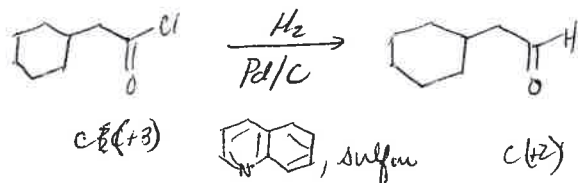
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How stop such reductions at C(+2) ox state?

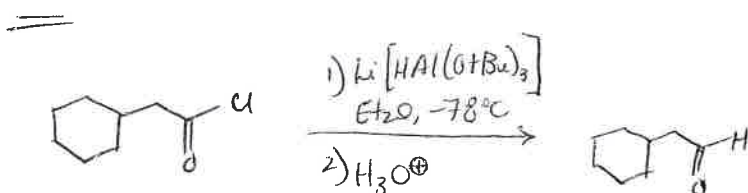
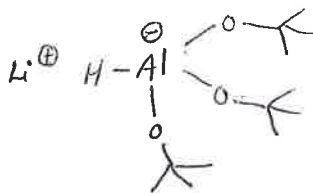
- start w/ ~~more~~ a more reactive carb. acid derivative, an acid chloride
- also, use a highly specialized reducing agent

1) Rosenmund Reduction - poisoned catalyst

ex:



2) Bulky derivative of $\text{LiAlH}_4 \rightarrow \text{Li}[\text{HAl}(\text{OtBu})_3]$



Reactions of carboxylic acid derivatives w/ organometallic reagents

"carbanion-like"

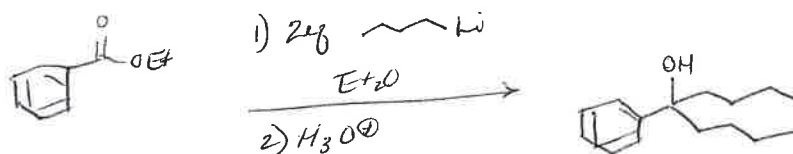
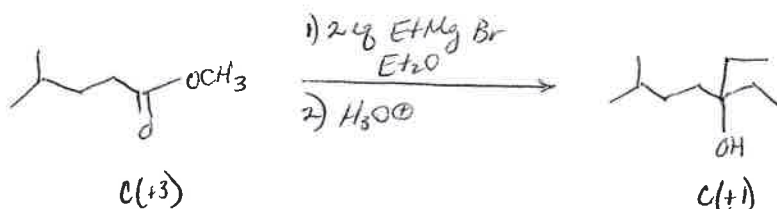
* form C-C bonds

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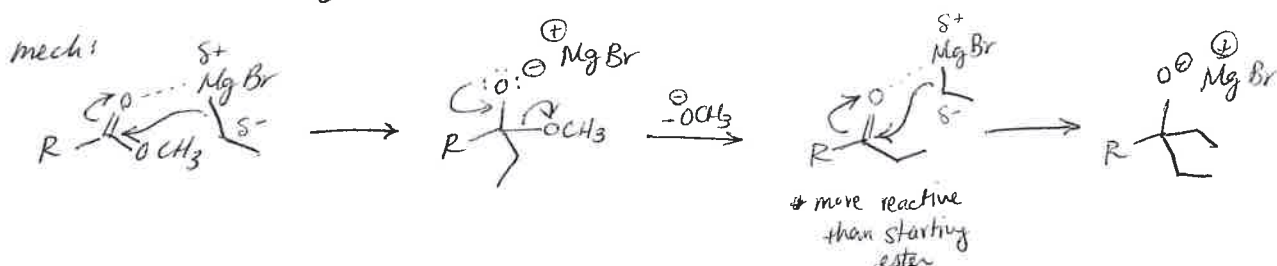
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1) Ester ; Grignard Reagents or organolithium reagents

ex:

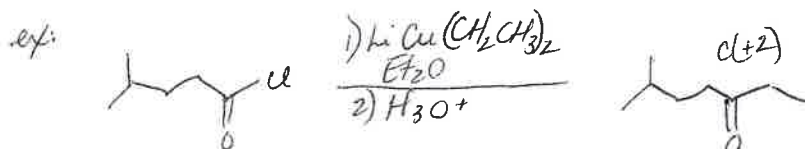


• cannot stop rxns @ C(+2) stage b/c ketones more reactive than starting ester



• how stop at ketone stage C(+2)?

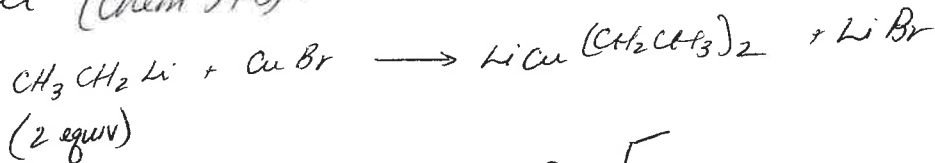
• acid chloride ; lithium diorganocuprate



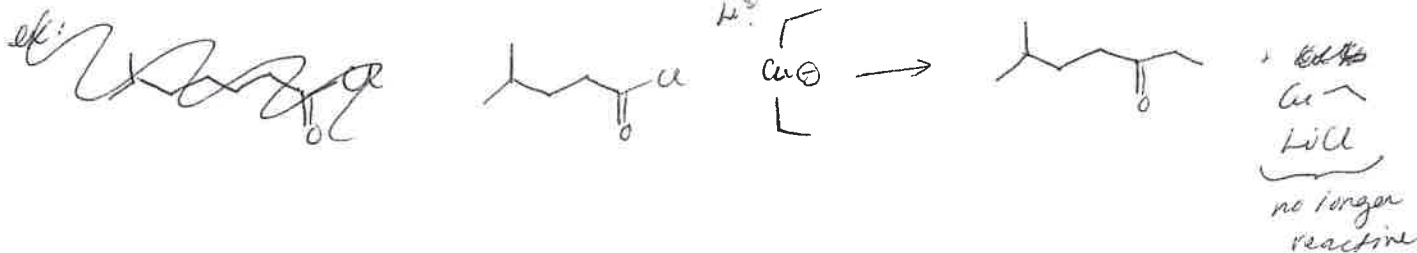
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Recall (chem 343):



key reaction:

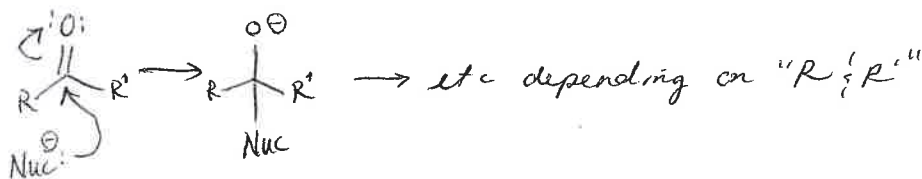


Read 21.11 - Synthesis of Carboxylic Acid Derivatives

Ch 22 - Enols; Enolates

Rec Probs - 1-9, 11-23, 25-31, 35, 36, 37a, 38, 40-44, 46-49, 51-95

Past Theme:



New Theme: "α-acidity"

