

## RECOMMENDED STUDY HABITS

1. Study regularly and often (every day, if possible). This course covers a large amount of material, and many of the concepts are difficult to master. Students are most likely to succeed if they are methodical in their study habits.

2. Recopy your class notes within 24 hours of the lecture. Many important points slip by during lecture before you can record them, but these points remain in your short-term memory. Recopying gives you the opportunity to set down the full story, and cements your grasp of concepts presented in the lecture. (This technique is not at all obvious; I learned of it from a professor when I was an undergraduate, and it changed my life.)

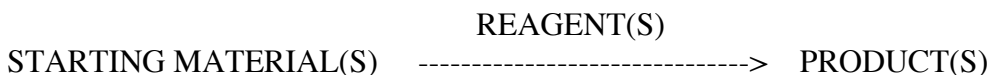
3. Read each chapter completely (unless sections are explicitly omitted). There is not time for all important material to be covered in lecture, and some key points will be left for the text to explain.

4. **Doing problems in the book and on sample exams is the most important element of studying organic chemistry. Problems must be approached in the proper way for students to derive full benefit.** Write out the answers to all recommended problems **before** you look at the printed solutions. It is seductively easy to look at a problem, think for a moment, look at the printed solution, and then tell yourself, "oh yes, I knew that." Each problem helps anchor some concept in a student's mind. Every time one looks at the answer before writing out the answer to a problem, most of what that problem could have taught is lost.

Student's can gauge their progress in the course by whether they can do the book problems without looking at the answers. Use your performance on the problems to identify the concepts and facts that you have not mastered. If one cannot do the problems, one probably can't succeed on the exams. The more one practices, the greater his or her facility will become. A student cannot do too many problems.

5. Organic Chemistry is a difficult subject because there is a great deal of information that must be committed to memory, and there are broad concepts and principles to be mastered. For the material that must be committed to memory, passive studying (i.e., multiple rereadings of text and lecture notes) has only limited effectiveness. Active studying is preferable; active study techniques for two important types of knowledge are explained below. (Students are encouraged to devise their own active study strategies for other types of knowledge important in this course.)

(i) New reactions. General form of information:



Study approach: two sets of flash cards.

(a) "Question side" has starting material(s) and reagent(s), "answer side" has product(s).

(b) "Question side" has starting material(s) and product(s), "answer side" has reagents(s).

Ideally you already have a set of reaction flash cards for material covered in Chemistry 343. If not, making those flash cards as Chemistry 345 begins is a good way to review prior material. Mastering the reactions in this way is crucial for solving "synthesis" problems, which require students to devise a series of reactions that convert a simple starting material into a more complex product.

(ii) Reaction mechanisms. Study approach: maintain a list of the important mechanisms discussed during lecture, and periodically practice drawing out a complete version of each important mechanism *without looking at your notes*. "Drawing out a mechanism" means proceeding from starting material to product, drawing each intermediate, and showing the flow of electrons (with curved arrows) that occurs in each elementary step of the mechanism. Many mechanisms are also discussed in terms of energy diagrams (energy on vertical scale, reaction coordinate on horizontal scale), in which starting material, product and intermediates are concave up points (minima on the curve), and transition states are concave down points (maxima on the curve). Energy diagrams and "curved arrow" mechanisms are complementary methods of representing a reaction.

6. Make sure that your command of the material from Chemistry 343 is strong; review that material if necessary. Students will need to make regular use of concepts and specific information from Chemistry 343. "Concepts" include stereochemistry, mechanism, bonding and resonance. "Specific information" includes the conditions for particular organic reactions (e.g., to convert an alkene to an alkane, or an alcohol to an ether) and conformational analysis.

7. General Perspective. Each student should be mindful of two main goals in this course: (1) to develop a solid understanding of the foundations of organic chemistry, and (2) to learn how to master a challenging intellectual discipline that requires both understanding of a complex conceptual framework and memorization of specific facts. Intrinsic interest in organic chemistry (goal #1) varies widely among the students who take this course, but every student should be attracted to goal #2. If you can learn how to master organic chemistry, then you will have developed intellectual tools that can be applied to mastering many other subject areas. Use this course to enhance your learning skills; the learning strategies that work best for you may differ from those that are most successful for others.

**(Recommended: Review this summary every two weeks.)**