## **Chemistry 103 – Course Schedule**

Week	Day	Date	Topic	Lab	Readings	Reading ?s	PS # Due*	Quiz #*
1	T	9/6	Matter & Measurement	Citizenship in lab	Ch. 1	1		
	R	9/8	Atoms, Molecules and Ions		Ch. 2			
2	Т	9/13	Atomic theory	A Solns, density & graphing B Lake study	Ch. 6 p. 207-218	2	1	1
	R	9/15	Quantum numbers and orbitals		Ch. 6 p. 219-229			
3	Т	9/20	Electron configurations & periodic trends	A Lake study B Solns, density & graphing	Ch. 6/7 p. 229-264 Ch. 8 p. 294-296	3	2	2
	R	9/22	Periodic trends, cont.		review Ch. 7			
4	Т	9/27	Bond types and Lewis structures	A Rxn of zinc and iodine B No lab	Ch. 8 p. 289-293 296-309	EXAM 1 – Mon 9/26 @ 7:15 PM		
	R	9/29	Lewis structures, cont.		review Ch. 8 p. 309-314			
5	Т	10/4	VSEPR theory and hybrid orbitals	A VSEPR tutorial     B Rxn of zinc and iodine	Ch. 9 p. 331-342 345-351	4	3	3
	R	10/6	Molecular polarity and π bonds		Ch. 9 p. 343-344 351-357			
6	Т	10/11	Intermolecular forces	A Light, color & solns B VSEPR tutorial	Ch. 11 p. 425-436	5	4	4
	R	10/13	Intermolecular forces, cont.		review Ch. 11			
7	Т	10/18	Stoichiometry	A Rxn types & chem. logic B light, color & solns	Ch. 3 p. 77-95	EXAM 2 – Mon 10/17 @ 7:15 PM		
	R	10/20	Stoichiometry, cont.		Ch. 3 p. 96-99			
8	Т	10/25	Limiting reagents and reaction yields	A Synthesis of an Alum B Rxn types & chem. logic	Ch. 3 p. 99-103	6	5	5
	R	10/27	Reactions in aqueous solutions		Ch. 4 p. 115-127			
9	Т	11/1	Neutralization and redox reactions	A No lab B Synthesis of an Alum	Ch. 4 p. 127-138	7	6	6
	R	11/3	Concentration, dilutions, titrations		Ch. 4 p. 139-148			
10	Т	11/8	Thermodynamics	A No lab B Solution calorimetry	Ch. 5 p. 159-169	EXAM 3 – I	Mon 11/7 @ 7	7:15 PM
	R	11/10	Enthalpy and calorimetry		Ch. 5 p. 169-180			
11	Т	11/15	Hess's Law and Heats of formation	A Solution calorimetry B No lab	Ch. 5 p. 181-187	8	7	7
	R	11/17	Bond enthalpies		Ch. 8 p. 315-321			
12	Т	11/22	Foods, fuels, current events in chemistry	_	Ch. 5 p. 188-194	EXAM 4 – N	lon 11/21 @	5:30 PM
	R	11/24	Happy Thanksgiving!					
13	Т	11/29	Gas laws	A No lab B Project lab	Ch. 10 p. 383-399	9	8	8
	R	12/1	Gases cont.		Ch. 10 p. 399-413			
14	Т	12/6	Liquids, phase changes, heating curves	A Project lab B No lab	Ch. 11 p. 437-448	10	9	9
	R	12/8	Solids – metals		Ch. 11 p. 448-452			
					Ch. 12 p. 463-476			
15	Т	12/13	Solids – ionic and network covalent	Window on the Solid State	Ch. 12 p. 481-487	11	10	10
	R	12/15	Solids – band theory		Ch. 12 p. 476-481 487-490			

<sup>\*</sup>Problem sets (PS) are due at the beginning of your Wednesday discussion section; quizzes are also administered at that time.