

PRINT NEATLY

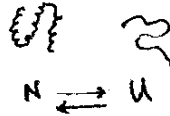
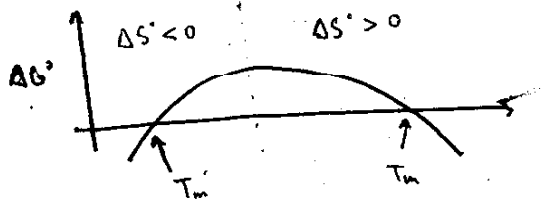
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DO NOT STAPLE

Course S6S/66S Lecture Number _____ Date 4/8/03

Lecturer Dr. Silvia Caragano Note Taker Eric Fulmer

Stability Curves



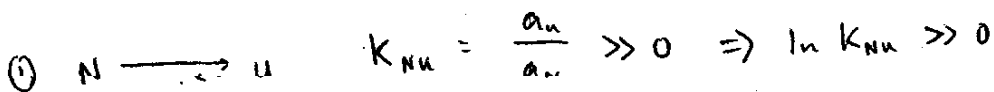
N ≡ Native State
U ≡ Unfolded State

$$\Delta G^\circ = \Delta H_m^\circ \left(1 - \frac{T}{T_m}\right) + \Delta C_p \left(T - T_m - T \ln\left(\frac{T}{T_m}\right)\right)$$

$$\Delta G^\circ = G_{\text{final}}^\circ - G_{\text{initial}}^\circ = G_u^\circ - G_n^\circ$$

Some Simple Power Tricks to Understand Equilibrium

$$\Delta G^\circ = -RT \ln K_{eq}$$



$$\boxed{\Delta G^\circ \ll 0}$$

In general, for right shifted ~~of~~ equilibria
(|U| > |N|) ⇒ $\boxed{\Delta G^\circ < 0}$

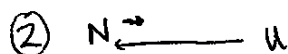
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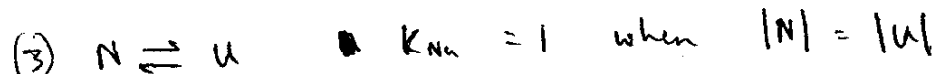
Course 565/665 Lecture Number _____ Date 4/8/03

Lecturer Cavagnero Note Taker Fulmer

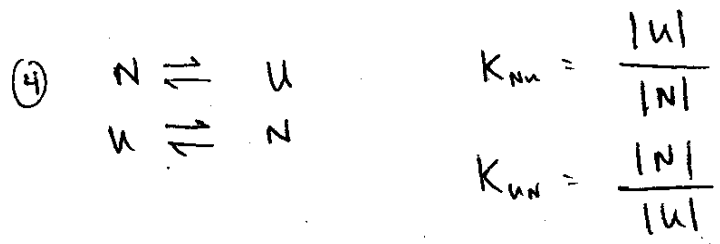


$K_{NU} \ll 1$, $\ln K_{NU} \ll 0$, $\Delta G^\circ \gg 0$

In general, for left-shifted equilibria, $\Delta G^\circ > 0$

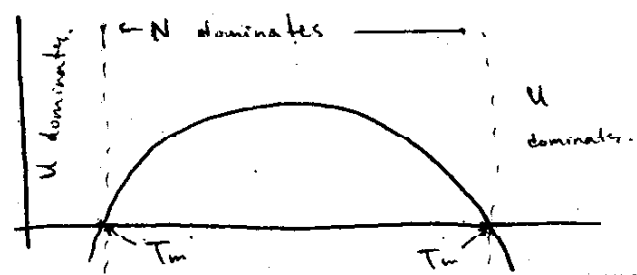


$\ln K_{NU} = 0 \Rightarrow \Delta G^\circ = 0$



$K_{UN} = \frac{1}{K_{NU}}$, $\ln K_{UN} = -\ln K_{NU}$

$\Delta G_{UN}^\circ = -\Delta G_{NU}^\circ$



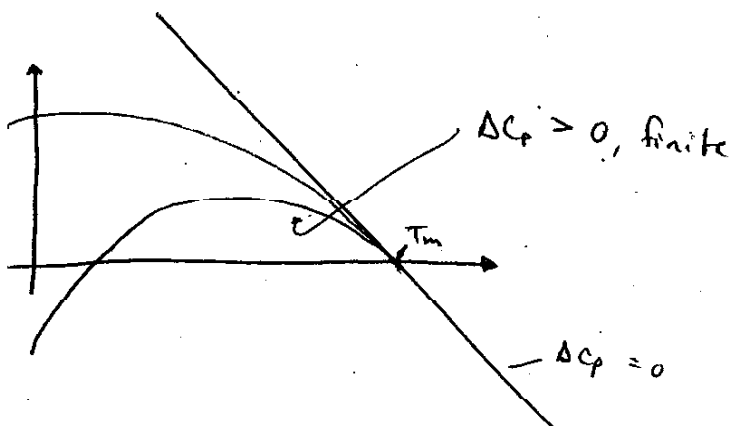
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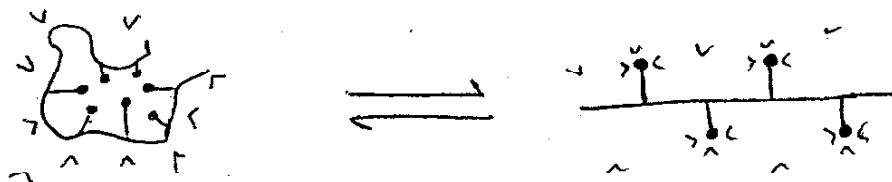
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Course S65/665 Lecture Number _____ Date 4/8/03

Lecturer Cavagnaro Note Taker Fulmer



$C_{p,n} > C_{p,d}$ Cp's are not equal.



"Iceberg Effect": High ordering of H_2O molecules around hydrophobic residues in an effort to solvate these groups. High entropic cost.