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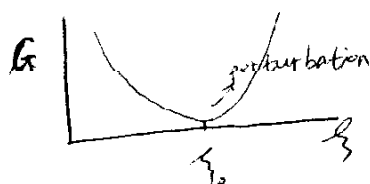
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Course 565/665 Lecturer Prof. Cavagnero
 Day 4.9.04 Date 9:55 am
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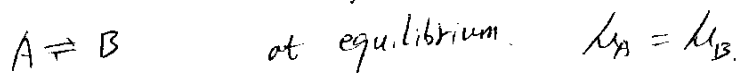
$$N_B = N\xi \quad \xi = \frac{N_B}{N} = \frac{N_B}{N_A + N_B} \quad ; \quad \text{in the beginning of rxn. } \xi = 0$$

$$\text{at equilibrium: } \xi_0 = \frac{N_B}{N}$$



$$\text{Towards equilibrium: } dG = (\mu_B - \mu_A) d\xi < 0$$

at constant p , how does K depend on T ?



$$\mu_A^\circ + RT \ln P_A = \mu_B^\circ + RT \ln P_B$$

$$\ln K_p = \ln \frac{P_B}{P_A} = \frac{-(\mu_B^\circ - \mu_A^\circ)}{RT} = -\frac{\Delta \mu^\circ}{RT}$$

$$\Delta \mu^\circ = -RT \ln K_p$$

$$\Delta \mu^\circ = \Delta H_{N_2}^\circ - T \Delta S_{N_2}^\circ$$

$$\Delta H_{N_2}^\circ \equiv \frac{\Delta H^\circ}{N_2}$$

$$\Delta S_{N_2}^\circ \equiv \frac{\Delta S^\circ}{N_2}$$



$$\left(\frac{\partial \ln K_p}{\partial T} \right)_P = - \frac{\partial \left(\frac{\Delta \mu^\circ}{RT} \right)}{\partial T} \Bigg|_P = - \frac{\partial}{\partial T} \left[\frac{\Delta H_{N_2}^\circ - T \Delta S_{N_2}^\circ}{RT} \right]_P$$

If $\Delta H_{N_2}^\circ$, $\Delta S_{N_2}^\circ$ do not depend on T :

write on this side only - do not double side for genchem office

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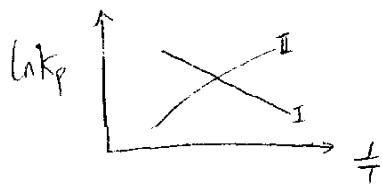
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$$\left(\frac{\partial \ln k_p}{\partial T} \right)_p = \frac{\Delta H_{rxn}^\circ}{kT^2} = \frac{\Delta H^\circ}{RT^2}$$

Recall $\left\{ \begin{array}{l} \textcircled{1} R = 8.314 \text{ J/K} \\ \textcircled{2} d\left(\frac{1}{T}\right) = -\frac{1}{T^2} dT \end{array} \right.$

$$\left(\frac{\partial \ln k_p}{\partial \frac{1}{T}} \right)_p = -\frac{\Delta H_{rxn}^\circ}{k} = -\frac{\Delta H^\circ}{R}$$

Van't Hoff equation



2 different situations:

I. $H_B^\circ > H_A^\circ \Rightarrow \Delta H^\circ > 0$ endothermic

II. $H_B^\circ < H_A^\circ \Rightarrow \Delta H^\circ < 0$ exothermic

recall: $H = q \Rightarrow$ intrinsic heat or energy (?)

Van't Hoff plot is linear for a lot of small molecule chem rxn.

(ex) know k_p at T_1 , calculate k_p at T_2 .

$$d \ln k_p = -\frac{\Delta H^\circ}{R} d\left(\frac{1}{T}\right)$$

$$\int_{\ln k_{p(T_1)}}^{\ln k_{p(T_2)}} d \ln k_p = -\int_{\frac{1}{T_1}}^{\frac{1}{T_2}} \frac{\Delta H^\circ}{R} d\left(\frac{1}{T}\right)$$

$$\ln k_{p(T_2)} - \ln k_{p(T_1)} = +\frac{\Delta H^\circ}{R} \left(\frac{1}{T_1} - \frac{1}{T_2} \right)$$