

1. A) Doolittle equation, $\eta = A e^{B(\frac{1}{f} - 1)}$

$$\ln \eta(T) = \ln A + B\left(\frac{1}{f(T)} - 1\right)$$

$$\ln \eta(T_0) = \ln A + B\left(\frac{1}{f(T_0)} - 1\right)$$

$$\ln \left[\frac{\eta(T)}{\eta(T_0)} \right] = B \left[\frac{1}{f(T)} - \frac{1}{f(T_0)} \right] \xrightarrow{f(T) = f(T_0) + \alpha_f(T - T_0)} B \left[\frac{1}{f(T_0) + \alpha_f(T - T_0)} - \frac{1}{f(T_0)} \right]$$

$$= B \left\{ \frac{f(T_0) - f(T_0) - \alpha_f(T - T_0)}{f(T_0)[f(T_0) + \alpha_f(T - T_0)]} \right\} = - \frac{[B/f(T_0)] \cdot (T - T_0)}{\frac{f(T_0)}{\alpha_f} + (T - T_0)}$$

$$\therefore \log \frac{\eta(T)}{\eta(T_0)} = - \frac{\frac{B}{2.303 f(T_0)} \cdot (T - T_0)}{\frac{f(T_0)}{\alpha_f} + (T - T_0)} = - \frac{C_1^0 (T - T_0)}{C_2^0 + (T - T_0)}, \quad C_1^0 = \frac{B}{2.303 f(T_0)}$$

$$C_2^0 = \frac{f(T_0)}{\alpha_f}$$

Hence, $f(T_1) = f(T_0) + \alpha_f(T_1 - T_0)$

$$C_1' = \frac{B}{2.303 f(T_1)} = \frac{B}{2.303 [f(T_0) + \alpha_f(T_1 - T_0)]} = \frac{B/2.303 \cdot f(T_0)}{1 + \frac{\alpha_f(T_1 - T_0)}{f(T_0)}}$$

$$C_2' = \frac{f(T_1)}{\alpha_f} = \frac{f(T_0) + \alpha_f(T_1 - T_0)}{\alpha_f} = \frac{f(T_0)}{\alpha_f} + (T_1 - T_0)$$

$$= \frac{C_1^0 f(T_0)}{1 + \frac{(T_1 - T_0)}{C_2^0}} = \frac{C_1^0 C_2^0}{C_2^0 + (T_1 - T_0)}$$

$$C_2^0 = C_2' + (T_0 - T_1)$$

$$C_1^0 C_2^0 = \frac{B}{2.303 \alpha_f} = C_1' C_2', \quad C_1^0 = \frac{C_1' C_2'}{C_2^0} = \frac{C_1' C_2'}{C_2' + T_0 - T_1}$$

B) VFT equation can be written as $\log \frac{\eta(T)}{\eta(T_0)} = B' \left(\frac{1}{T - T_\infty} - \frac{1}{T_0 - T_\infty} \right)$

$$B' \left(\frac{1}{T - T_\infty} - \frac{1}{T_0 - T_\infty} \right) = B' \left(\frac{T_0 - T_\infty - T + T_\infty}{(T - T_\infty)(T_0 - T_\infty)} \right) =$$

$$= - \frac{B'(T - T_0)}{(T_0 - T_\infty)(T - T_\infty)} = - \frac{B' \cdot (T - T_0)}{(T_0 - T_\infty) \cdot (T - T_\infty)} = - \frac{B' (T - T_0)}{(T_0 - T_\infty) + (T - T_\infty)}$$

$$= - \frac{\alpha (T - T_0)}{B + (T - T_0)}, \quad \log \frac{\eta(T)}{\eta(T_0)} = - \frac{\alpha (T - T_0)}{B + (T - T_0)} \text{ which is WLF equation}$$

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$$c) \log \frac{\eta(255)}{\eta(205)} = - \frac{16.6(50)}{104.4 + 50} = -5.38$$

$$\eta(255) = 10^{12} \text{ Pa}\cdot\text{s} \cdot 10^{-5.38} = 4.21 \cdot 10^6 \text{ Pa}\cdot\text{s}$$

$$\eta(305) = 10^{12} \text{ Pa}\cdot\text{s} \cdot 10^{-8.12} = 7.56 \cdot 10^3 \text{ Pa}\cdot\text{s}$$

$$\eta(355) = 10^{12} \text{ Pa}\cdot\text{s} \cdot 10^{-9.79} = 1.63 \cdot 10^2 \text{ Pa}\cdot\text{s}$$

2. Under isobaric conditions, $\alpha dT = \frac{dL_0}{L_0}$, $\alpha \int_{T_0}^T dT = \int_{L_0^*}^{L_0} \frac{dL_0}{L_0}$

$$\alpha(T-T_0) = \ln \frac{L_0(T)}{L_0^*(T_0)}$$

$\uparrow$
 $\alpha \neq f(T)$

$$L_0(T) = L_0^*(T_0) e^{\alpha(T-T_0)}$$

$$\varepsilon = \frac{L}{L_0} - 1, \quad L = L_0(1+\varepsilon), \quad \left(\frac{\partial L}{\partial T}\right)_{P,\varepsilon} = (1+\varepsilon) \left(\frac{\partial L_0}{\partial T}\right)_{P,\varepsilon}$$

$$= (1+\varepsilon) L_0^*(T_0) \alpha e^{\alpha(T-T_0)}$$

$$= (1+\varepsilon) \alpha L_0(T) = \alpha L(T)$$

Since $df = \left(\frac{\partial f}{\partial T}\right)_{P,L} dT + \left(\frac{\partial f}{\partial L}\right)_{P,T} dL$

$$\left(\frac{\partial f}{\partial T}\right)_{P,\varepsilon} = \left(\frac{\partial f}{\partial T}\right)_{P,L} + \left(\frac{\partial f}{\partial L}\right)_{P,T} \underbrace{\left(\frac{\partial L}{\partial T}\right)_{P,\varepsilon}}_{\parallel \alpha L(T)}$$

$$\therefore \left(\frac{\partial f}{\partial T}\right)_{P,L} = \left(\frac{\partial f}{\partial T}\right)_{P,\varepsilon} - \alpha L(T) \left(\frac{\partial f}{\partial L}\right)_{P,T} \quad \text{Q.E.D.}$$

3. Equilibrium swelling condition is $\Delta\mu_i = \Delta\mu_{i,m} + \Delta\mu_{i,e} = 0$

$$\Delta\mu_{i,e} = \left(\frac{\partial \Delta A}{\partial n_i}\right)_{T,P} = \left\{ \frac{\partial}{\partial n_i} \left[\frac{3}{2} \frac{RT}{M_s} (\lambda^2 - 1) \right] \right\}_{T,P}$$

$\downarrow$
isotropic swelling, $\lambda_x = \lambda_y = \lambda_z = \lambda$

Note that $\lambda^3 = 1/\phi_2$ (Swelling ratio) $= 1 + n_i \bar{V}_i$

$$\lambda^2 = (1 + n_i \bar{V}_i)^{2/3}, \quad \left(\frac{\partial \lambda^2}{\partial n_i}\right)_{T,P} = \frac{2}{3} (1 + n_i \bar{V}_i)^{-1/3} \cdot \bar{V}_i = \frac{2}{3} \phi_2^{1/3} \bar{V}_i$$

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$$\Delta \mu_{1,c} = \frac{3}{2} \frac{SRT}{M_s} \cdot \frac{2}{3} \phi_2^{1/3} \bar{V}_1 = \frac{SRT}{M_s} \bar{V}_1 \phi_2^{1/3}$$

$$\Delta \mu_1 = RT \left[-\phi_2 - \frac{1}{2} \phi_2^2 - \dots + \phi_2 + \chi_1 \phi_2^2 \right] + \frac{SRT}{M_s} \bar{V}_1 \phi_2^{1/3} = 0$$

↑ When $\phi_2 \ll 1$ (large swell ratio limit)

$$\phi_2^2 \left(\chi_1 - \frac{1}{2} \right) + \frac{S \bar{V}_1}{M_s} \phi_2^{1/3} = 0$$

Equilibrium swell at large swelling limit

$$\frac{S \bar{V}_1}{M_s} = \left(\frac{1}{2} - \chi_1 \right) \phi_2^{5/3}, \text{ hence } G = \frac{SRT}{M_s} = \frac{RT}{\bar{V}_1} \left(\frac{1}{2} - \chi_1 \right) \phi_2^{5/3}$$

Q.E.D.

$$4. \quad \Delta \bar{S} = \int_{T_{ref}}^T \bar{C}_p d \ln T = \int_{T_{ref}}^T \frac{\bar{C}_p}{T} dT$$

T_1 : Tg of polymer 1 in pure state

T_2 : Tg of " 2 "

$\bar{S}_{total} = X_1 \bar{S}_1 + X_2 \bar{S}_2$, $\bar{C}_{pi}$ molar heat capacity of "i"

Melt state: $\bar{S}_{melt} = X_1 \bar{S}_{1,melt} + X_2 \bar{S}_{2,melt}$

$$= X_1 \left\{ \bar{S}_{1,melt}^0 + \int_{T_1}^{T_g} \bar{C}_{p1,melt} d \ln T \right\}$$

$$+ X_2 \left\{ \bar{S}_{2,melt}^0 + \int_{T_2}^{T_g} \bar{C}_{p2,melt} d \ln T \right\}$$

Glass state: $\bar{S}_{glass} = X_1 \bar{S}_{1,glass} + X_2 \bar{S}_{2,glass}$

$$= X_1 \left\{ \bar{S}_{1,glass}^0 + \int_{T_1}^{T_g} \bar{C}_{p1,glass} d \ln T \right\}$$

$$+ X_2 \left\{ \bar{S}_{2,glass}^0 + \int_{T_2}^{T_g} \bar{C}_{p2,glass} d \ln T \right\}$$

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 $\bar{S}_{\text{melt}} = \bar{S}_{\text{glass}}$ for each component

$$\bar{S}_{1,\text{melt}}^{\circ}(T_1) = \bar{S}_{1,\text{glass}}^{\circ}(T_1)$$

$$\bar{S}_{2,\text{melt}}^{\circ}(T_2) = \bar{S}_{2,\text{glass}}^{\circ}(T_2)$$

Also $\bar{S}_{\text{melt}}(T_g) = \bar{S}_{\text{glass}}(T_g)$

$$\begin{aligned} X_1 \bar{S}_{1,\text{melt}}^{\circ} + X_2 \bar{S}_{2,\text{melt}}^{\circ} + X_1 \int_{T_1}^{T_g} \bar{C}_{p1,\text{melt}} d\ln T + X_2 \int_{T_2}^{T_g} \bar{C}_{p2,\text{melt}} d\ln T \\ = X_1 \bar{S}_{1,\text{glass}}^{\circ} + X_2 \bar{S}_{2,\text{glass}}^{\circ} + X_1 \int_{T_1}^{T_g} \bar{C}_{p1,\text{glass}} d\ln T + X_2 \int_{T_2}^{T_g} \bar{C}_{p2,\text{glass}} d\ln T \\ X_1 \int_{T_1}^{T_g} (\bar{C}_{p1,\text{melt}} - \bar{C}_{p1,\text{glass}}) d\ln T + X_2 \int_{T_2}^{T_g} (\bar{C}_{p2,\text{melt}} - \bar{C}_{p2,\text{glass}}) d\ln T = 0 \end{aligned}$$

$$X_1 \Delta \bar{C}_{p1} (\ln T_g - \ln T_1) + X_2 \Delta \bar{C}_{p2} (\ln T_g - \ln T_2) = 0$$

$$n_1 \Delta \bar{C}_{p1} (\ln T_g - \ln T_1) + n_2 \Delta \bar{C}_{p2} (\ln T_g - \ln T_2) = 0$$

$$\left(\frac{m_1}{M_1}\right)(\Delta C_{p,1} \cdot M_1)(\ln T_g - \ln T_1) + \frac{m_2}{M_2}(\Delta C_{p,2} \cdot M_2)(\ln T_g - \ln T_2) = 0$$

↑ specific heat capacity difference

$$w_1(\Delta C_{p,1})(\ln T_g - \ln T_1) + w_2(\Delta C_{p,2})(\ln T_g - \ln T_2) = 0$$

↑ weight fraction $\frac{m_1}{m_1 + m_2}$

$$\ln T_g = \frac{w_1(\Delta C_{p,1}) \ln T_1 + w_2(\Delta C_{p,2}) \ln T_2}{w_1 \Delta C_{p,1} + w_2 \Delta C_{p,2}}$$

Q.E.D.

B) If $T_1 \approx T_2$ and $\Delta C_{p,1} = \Delta C_{p,2}$

$$\ln T_g = \frac{w_1 \ln T_1 + w_2 \ln T_2}{w_1 + w_2} \quad \text{since denominator} = 1 = w_1 + w_2$$

↑ $\Delta C_{p,1} = \Delta C_{p,2}$

$$w_1 \ln T_1 + w_2 \ln T_2 - (w_1 + w_2) \ln T_g = 0$$

$$w_1 \ln\left(\frac{T_1}{T_g}\right) + w_2 \ln\left(\frac{T_2}{T_g}\right) = 0$$

since $T_1 \approx T_2$, $T_1 \approx T_g \approx T_2$, $\frac{T_1}{T_g} \approx 1$, $\frac{T_2}{T_g} \approx 1$

$$\ln\left(\frac{T_1}{T_g}\right) \approx \frac{T_1}{T_g} - 1, \quad \ln\left(\frac{T_2}{T_g}\right) \approx \frac{T_2}{T_g} - 1, \quad \ln(1+x) = x - \frac{1}{2}x^2 + \frac{1}{3}x^3 - \dots$$

$$w_1 \frac{T_1}{T_g} - w_1 + w_2 \frac{T_2}{T_g} - w_2 = 0, \quad T_g = w_1 T_1 + w_2 T_2 \quad \text{Q.E.D.}$$