

Course Chem 345 Lecturer Gellman
Day Friday Date 1/20/17
Notes Taken By Anna Brezny Total # of Pages 5

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Recall: Use of spectroscopy to learn about molecular structure

- Example: IR — detect specific types of bonds
- Use spectroscopy data to test hypothesis about molecular structure or to formulate a hypothesis

Chapter 13: Nuclear Magnetic Resonance (NMR) spectroscopy

^1H NMR — signal for every type of H in molecule

^{13}C NMR — every type of ^{13}C in molecule

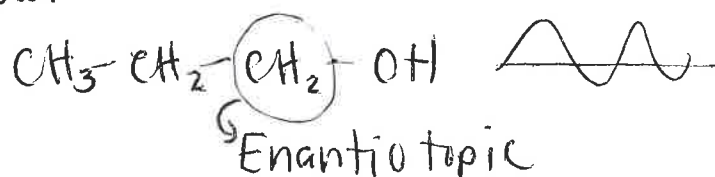
Rec. Problems: 3, 6-28, ~~31-47~~, 52, 53, 55-59

Equivalent H's vs. non-equivalent H's
→ All equivalent H's in a molecule contribute to the same NMR signal ("resonance").

Relationships among H's in a molecule:

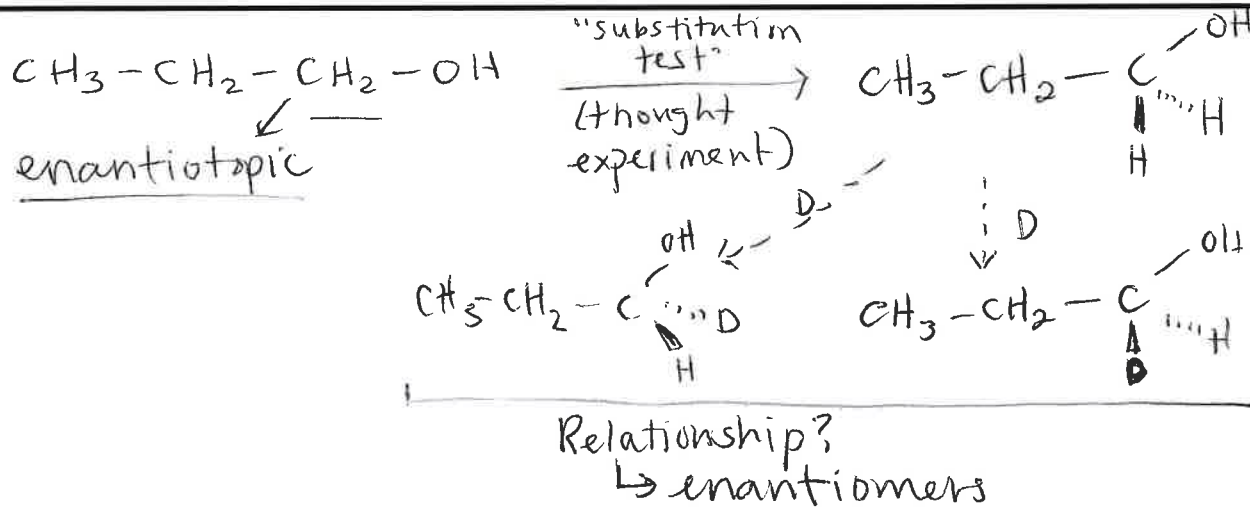
- Homotopic
 - Enantiotopic
 - Diastereotopic
- } Section 10.9

Consider:

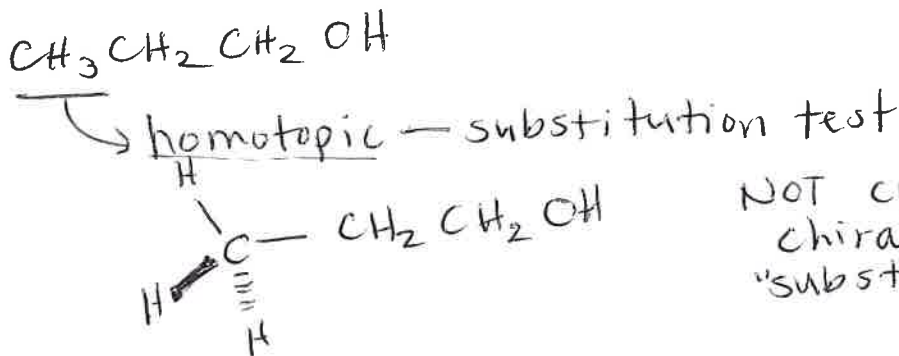


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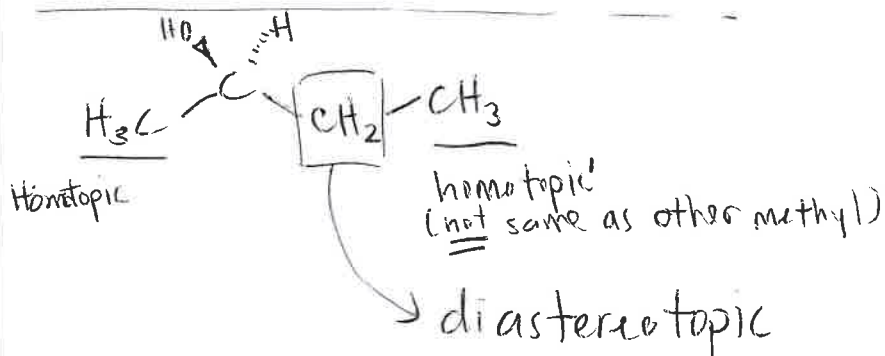
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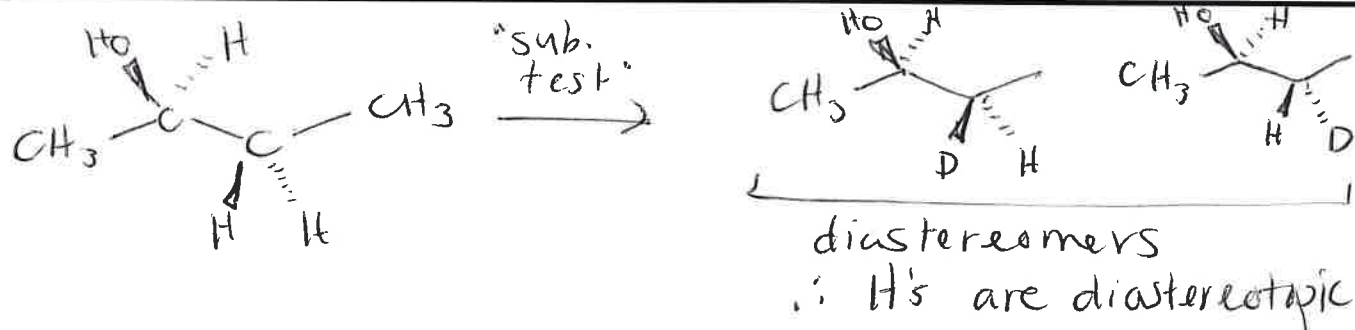
$\text{CH}_3 \text{ CH}_2 \text{ CH}_2 \text{ OH}$
 ↳ also enantiotopic
 prove this to yourself



NOT creating a chiral center w/ "substitution test"



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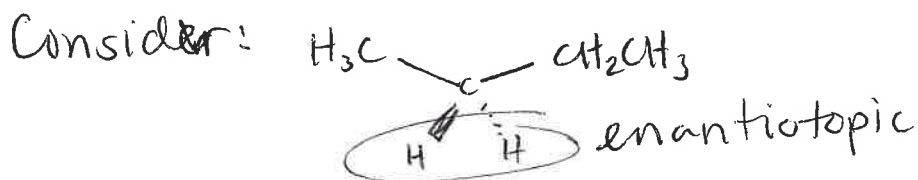


Equivalent vs. non-equiv. H's - EXAMPLES

- ① CH_3CH_3 All H's equivalent
 \Rightarrow single ^1H NMR resonance (all 6 H's)
- ② $\text{CH}_3\text{CH}_2\text{CH}_3$ All methyl H's equivalent to one another (6 total)
 Both CH_2 H's equivalent to one another (2 total)
 \Rightarrow 2 ^1H NMR resonances

Note: All 6 CH_3 H's are homotopic, homotopic H's are equivalent to one another

- ③ $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3$ 6x methyl H's - homotopic
 \therefore one resonance
 4x methylene (CH_2) H's - ~~are ... ?~~
 are ... ?
 let's see!




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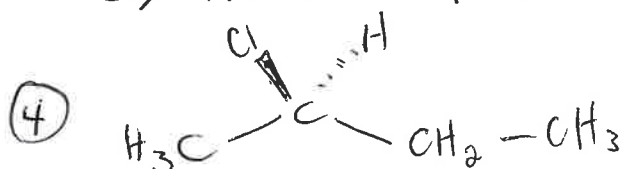
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enantiotopic H's are equivalent in terms of NMR in an achiral environment

Most solvents you are used to are achiral!

All 4 CH₂ H's in  are NMR equivalent

⇒ ¹H NMR spectrum of  has 2 resonances



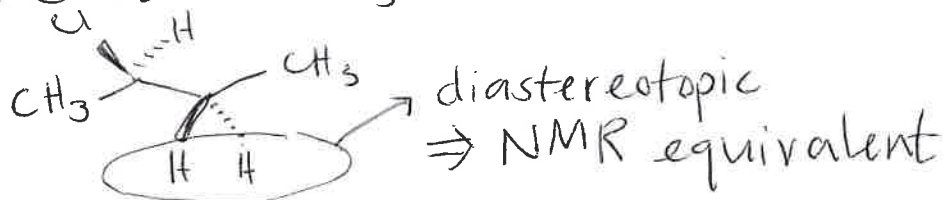
5 ¹H NMR ~~signals~~ signals

(i) H₃C @ C₁

(ii) H₃C @ C₄

(iii) H @ C₂ (bearing Cl)

(iv) ≡ (v)

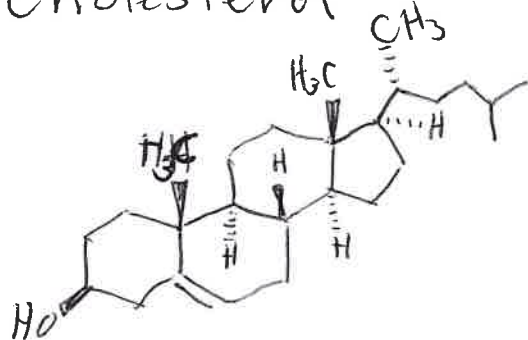


Sometimes diastereotopic H's will accidentally overlap. But purely coincidental

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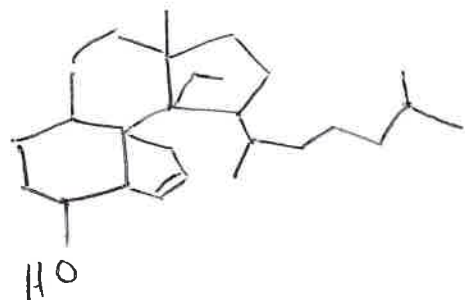
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Cholesterol



36 different
kinds of H's

Proposed structure (1928)



What ~~are~~ do we measure by NMR ?

→ Nuclear spin

remember that there are relationships ~~are~~
between charged entities & magnetic fields